

Controlled release micronutrient fertilizers for precision agriculture

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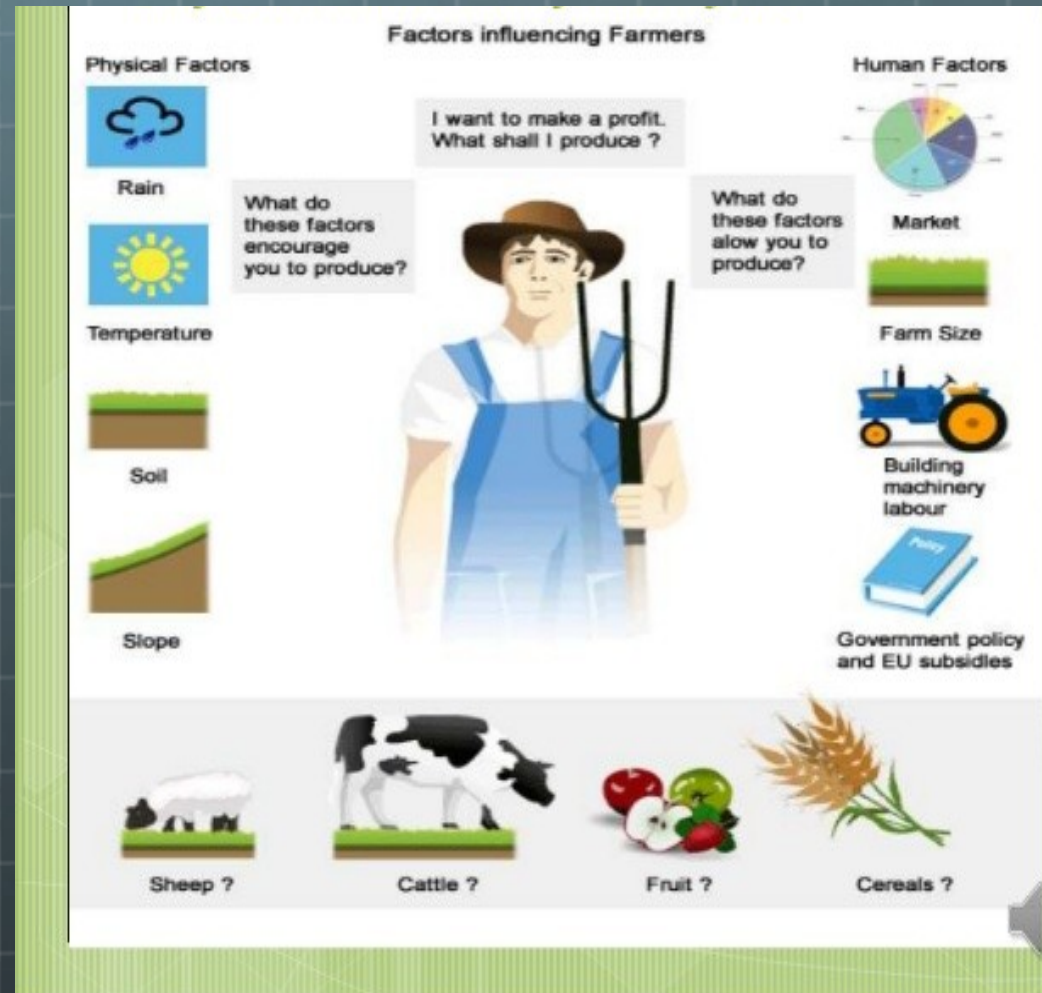
Project “Crop plants and natural products as a source of biologically active substances for the manufacture of cosmetics, pharmaceuticals and dietary supplements” (BIOSTRATEG2/298205/9/NCBR/2016) by The National Centre for Research and Development in Poland.

Precision agriculture

- 🌐 1980's: 'site-specific agriculture/crop management'.
- 🌐 The term appeared first in **1990**.
- 🌐 Aims to optimize field level management:
 - 🌐 **Crops** - matching farming practices with crop needs (fertilizer inputs)
 - 🌐 **Environmental protection** - reduces environmental risks and footprint of farming (limiting leaching of nutrients)
 - 🌐 **Economics** - competitiveness by more efficient practices (improved management of fertilizer use)

Definition

- PA utilizes innovative, **site-specific techniques** for management of spatial and temporal variability by affordable technologies to enhance:
 - output,**
 - efficiency**
 - and profitability**
- of agricultural production in **environmentally responsible** manner.
- Assessment of nutrient content of **soil and plants.**



- 🌐 **Fertilizers** are responsible for approximately **half** of the world's **crop** production - important for global food security.
- 🌐 Most fertilizers - **concentrated minerals** from **ore deposits**; exception - N.
- 🌐 Sufficient crop production depends on efficient fertilizers.
- 🌐 Integrated system.
- 🌐 Challenge - to provide:
 - 🌐 the right **nutrients**,
 - 🌐 at the right **rate**,
 - 🌐 at the right **time**
 - 🌐 and in the right **place**.

NUE - Nutrients Use Efficiency

- 🌐 **N** use efficiency **10-30 %** (urea),
- 🌐 **P** use efficiency **30-40 %** (single superphosphate)
- 🌐 **Micronutrients 30-50 %**
- 🌐 Only **40-50 %** of applied fertilizer is used by the plants.
- 🌐 **Losses** of fertilizer nutrients depend on soil processes:
 - 🌐 **microbiological** transformations
 - 🌐 **chemical** - exchange, precipitation, hydrolysis
 - 🌐 **physical** - leaching, runoff, volatilization

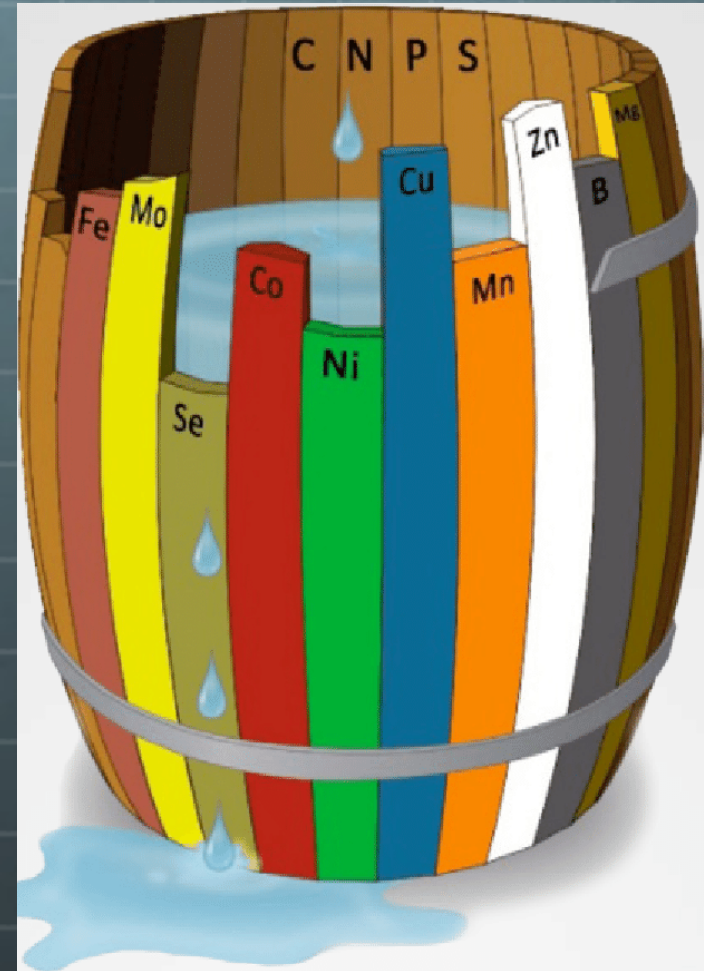
There are **17 essential nutrients** for crop growth.

C, H, O - air & water

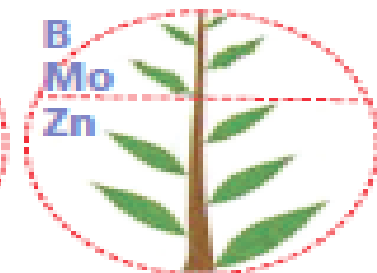
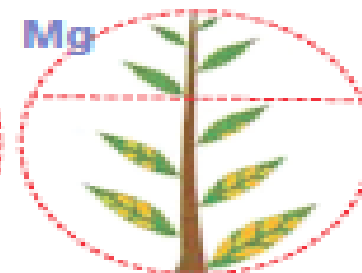
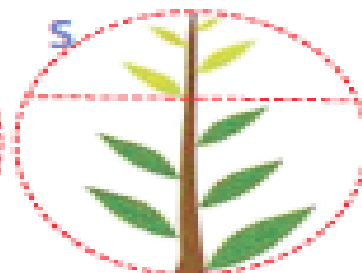
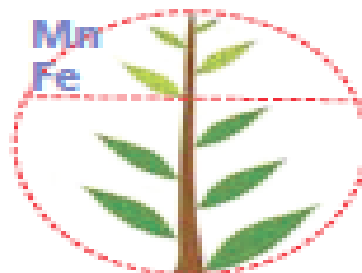
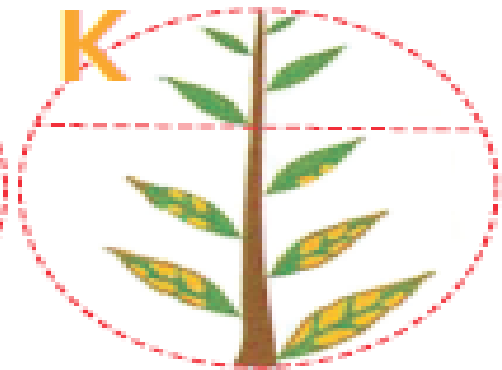
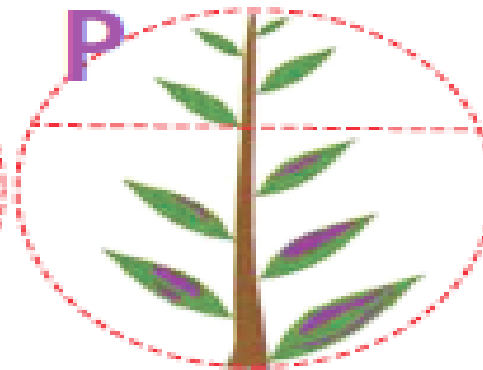
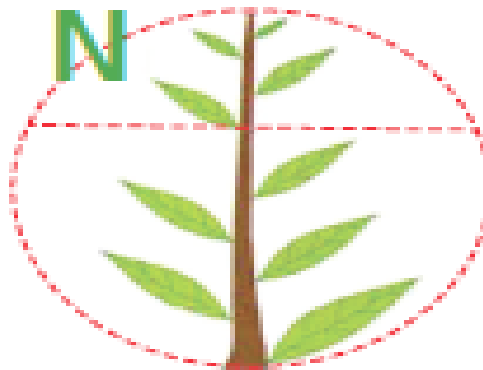
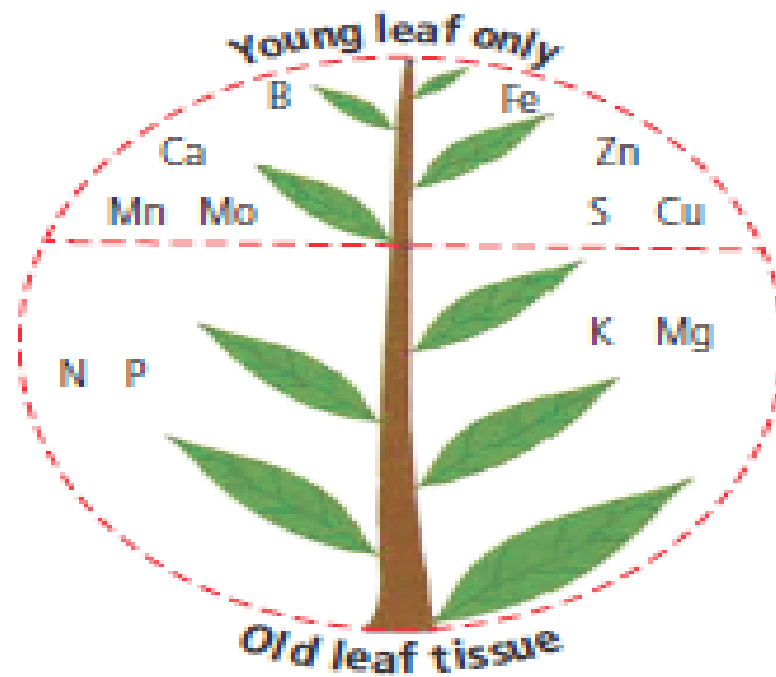
macronutrients—NPK

secondary nutrients—S, Ca, Mg

micronutrients—B, Fe, Mn, Zn, Cu, Mo, Cl, Ni - small amounts, catalysts in metabolic processes



Locations of nutrient deficiency symptoms on plants



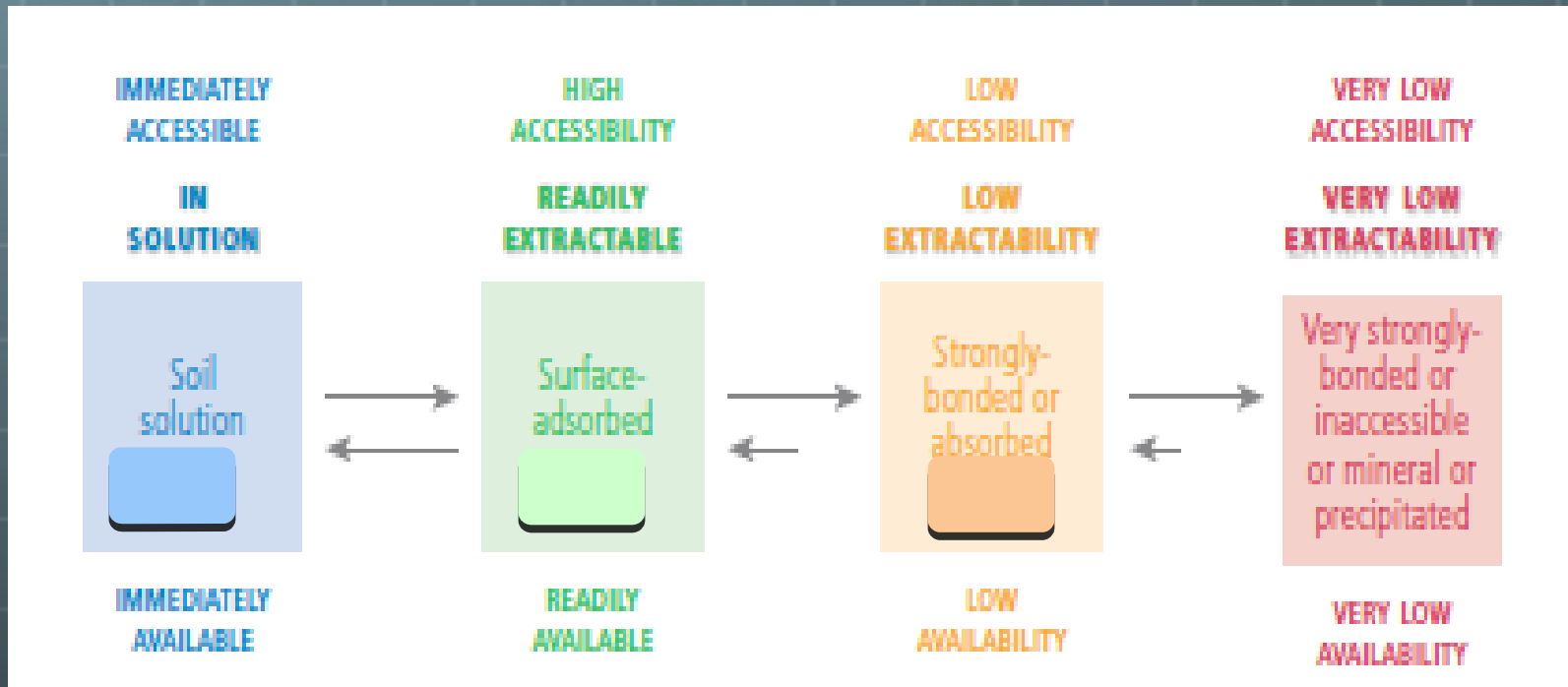
release micronutrient

fertilizers

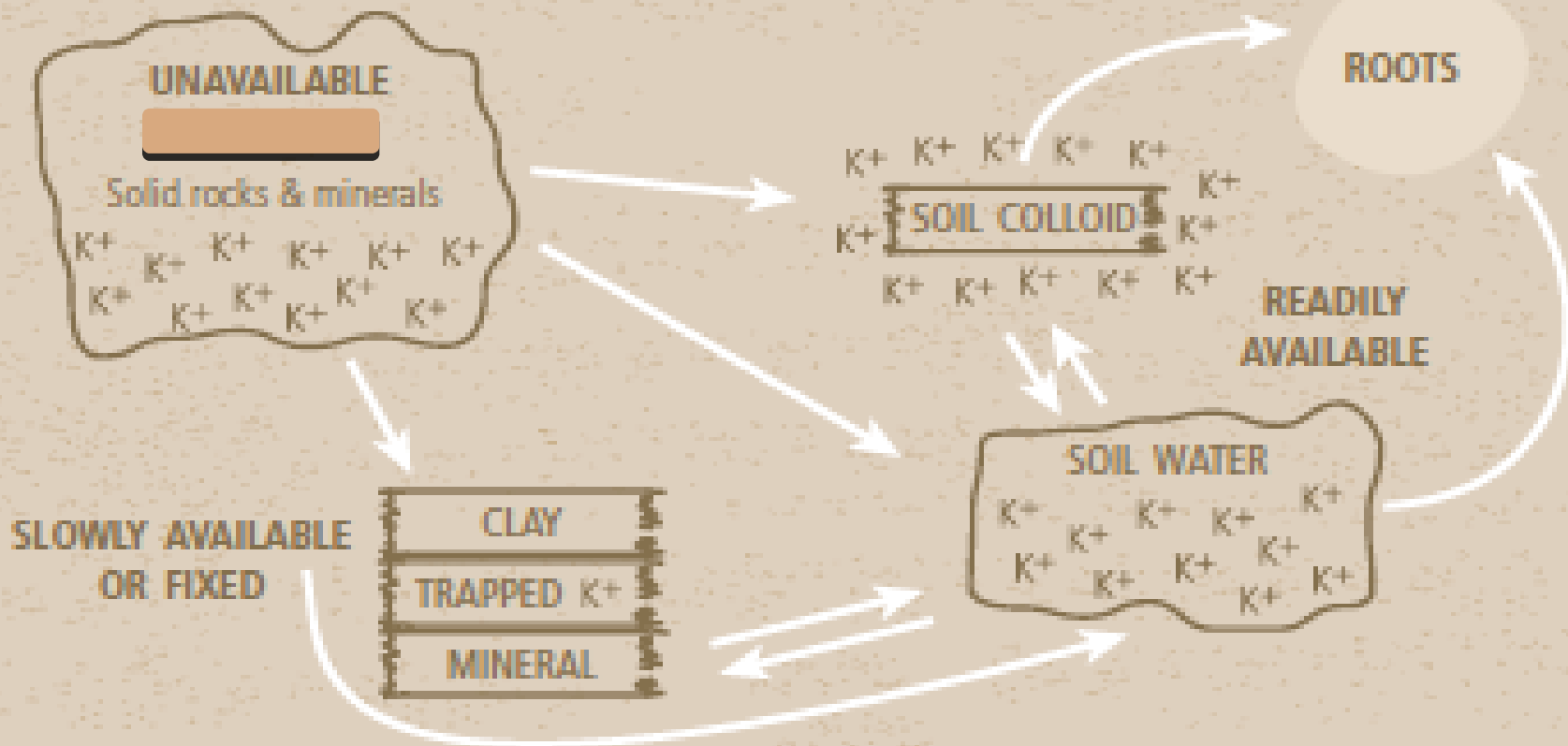
- 🌐 The research focuses on **increasing efficiency** of the **existing** fertilizers.
- 🌐 Doubling efficiency of the fertilizer means **reduction** of the **applied fertilizer by 50 %**
- 🌐 **Micronutrients** - essential components of enzymes/proteins important to increase crop yield and nutritional quality of food.
- 🌐 Conventional - bulk micronutrients - water soluble: **sulfates or chelated** forms, DTPA (diethylene triamine penta acetic acid), EDTA.

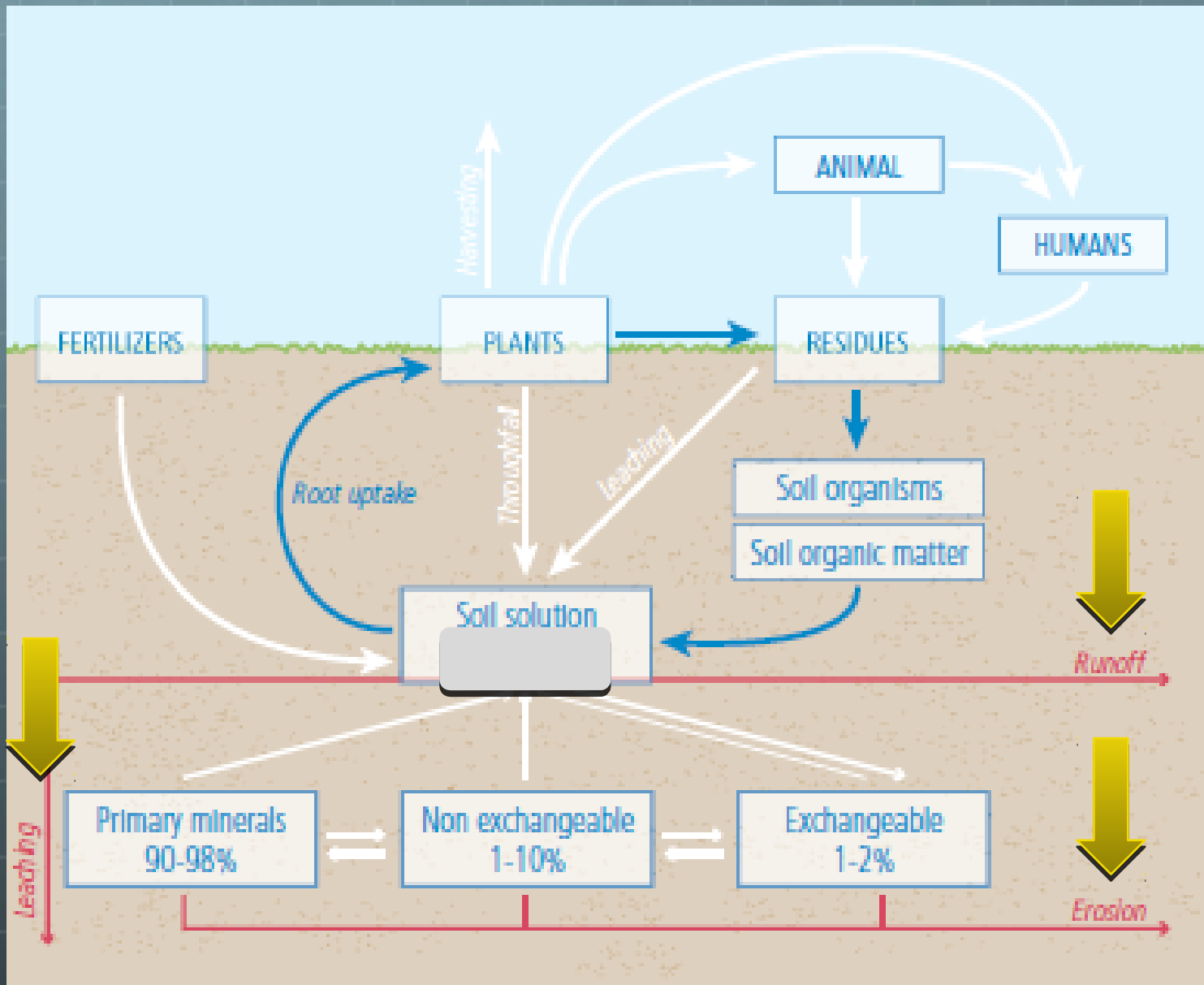
soil.

With conditions change (pH, concentration) - amounts in each pool change



- Cation nutrients in various **equilibrium** positions in soil.
- As cation is taken up by plants, the equilibrium **shifts** to release more cations into the soil solution
- **Plant roots** can decompose insoluble compounds by ion exchange with the root hairs or by **extracellular organic acid** that extract nutrients by **chelation**.



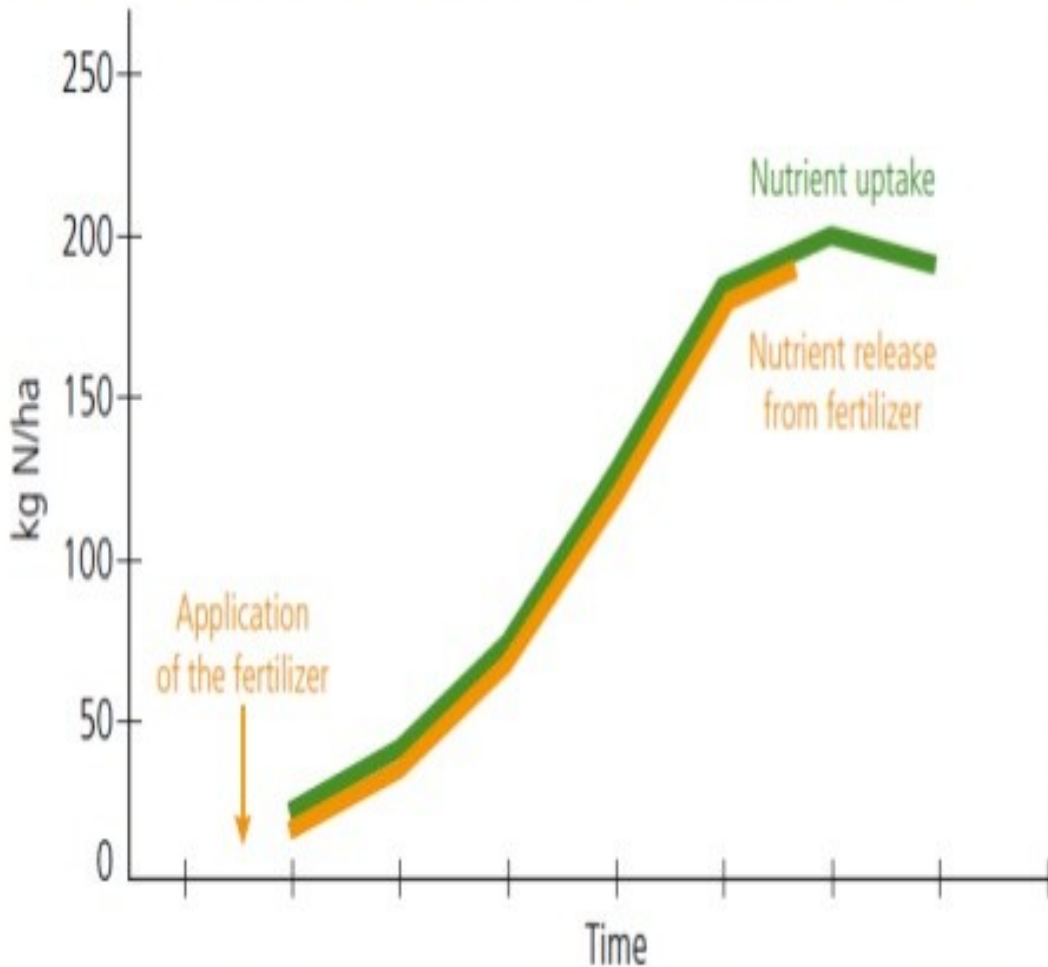


Optimal Plant Development - nutrients are precisely supplied according to specific plant needs



- 🌐 The goal: **optimal nutrition** throughout the growing season
- 🌐 **Granular soluble fertilizer:**
 - 🌐 dangerous **excess** at the beginning
 - 🌐 followed by **deficiency** towards the end of the growing season

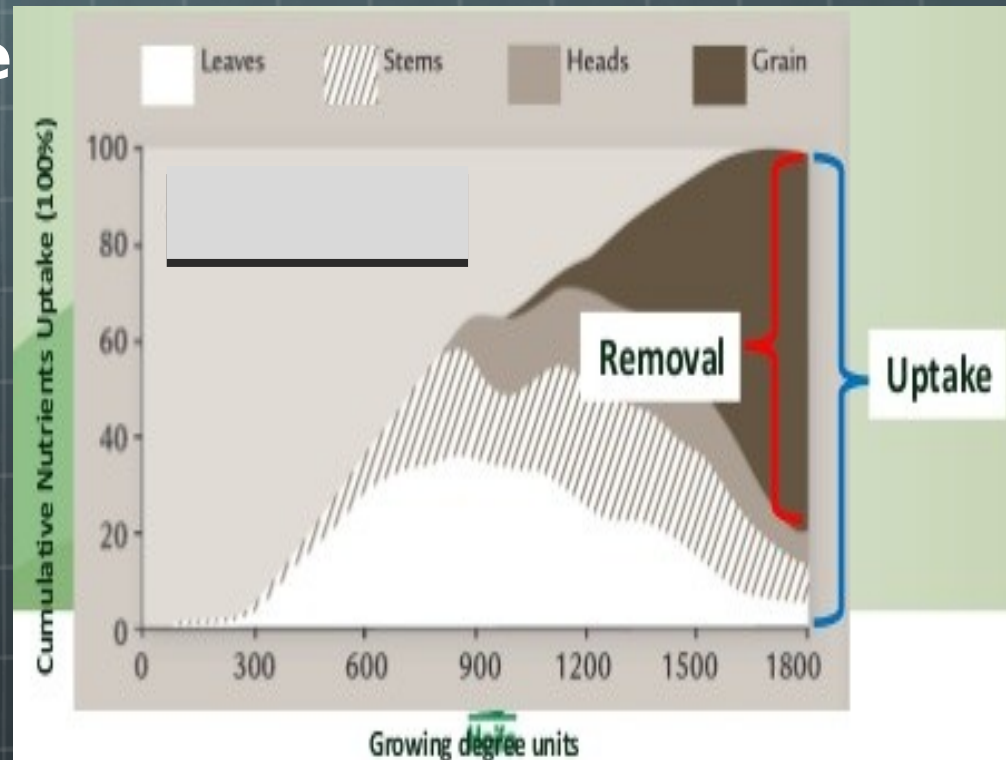
The nutrient release should be **synchronized** with plant nutrient requirements



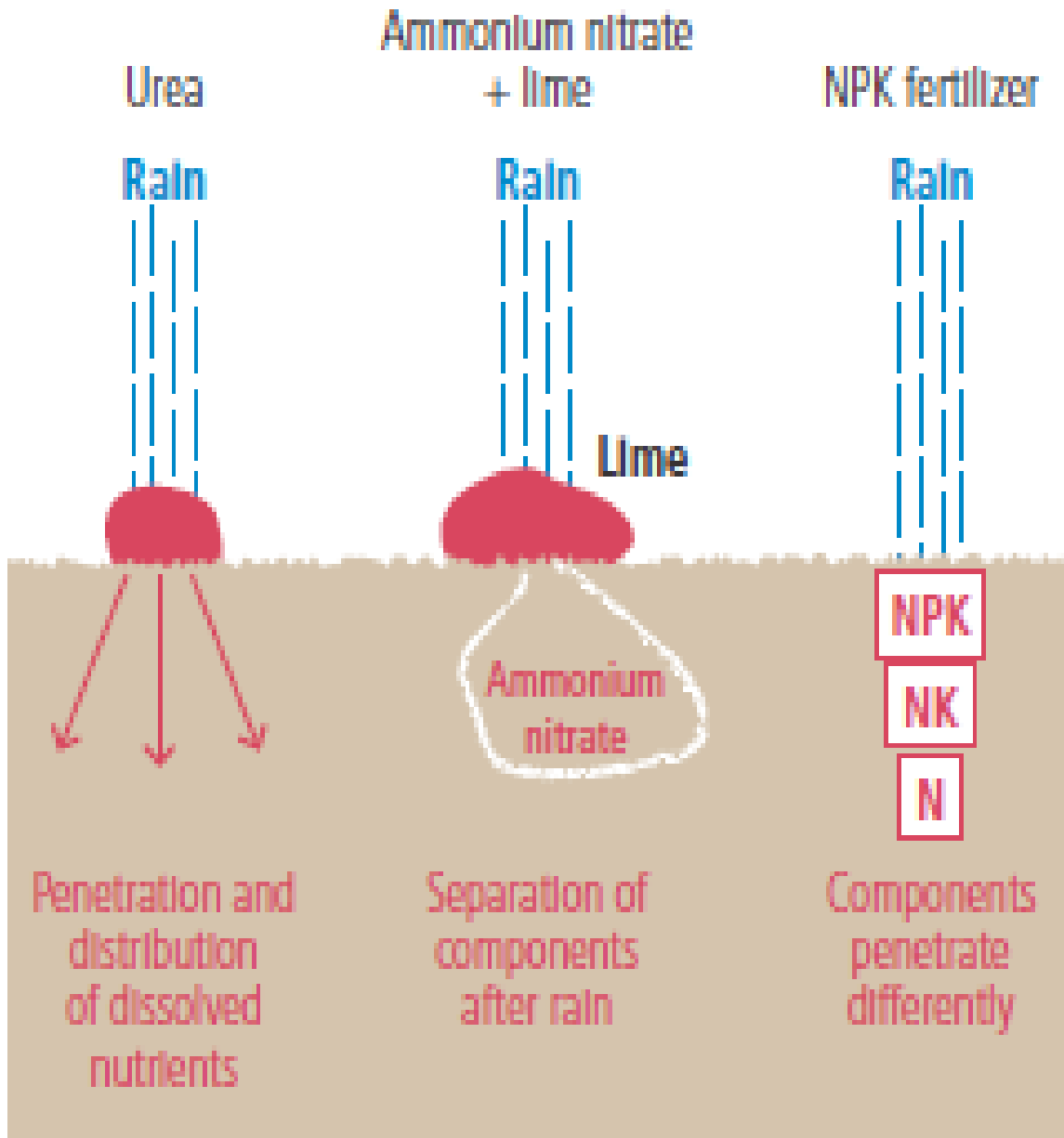
- 🌐 The nutrient release patterns should **match exactly** the nutrient uptake curve of the crop
- 🌐 Any excess nutrients shall remain in the soil for the next crop

NUTRIENTS FATE

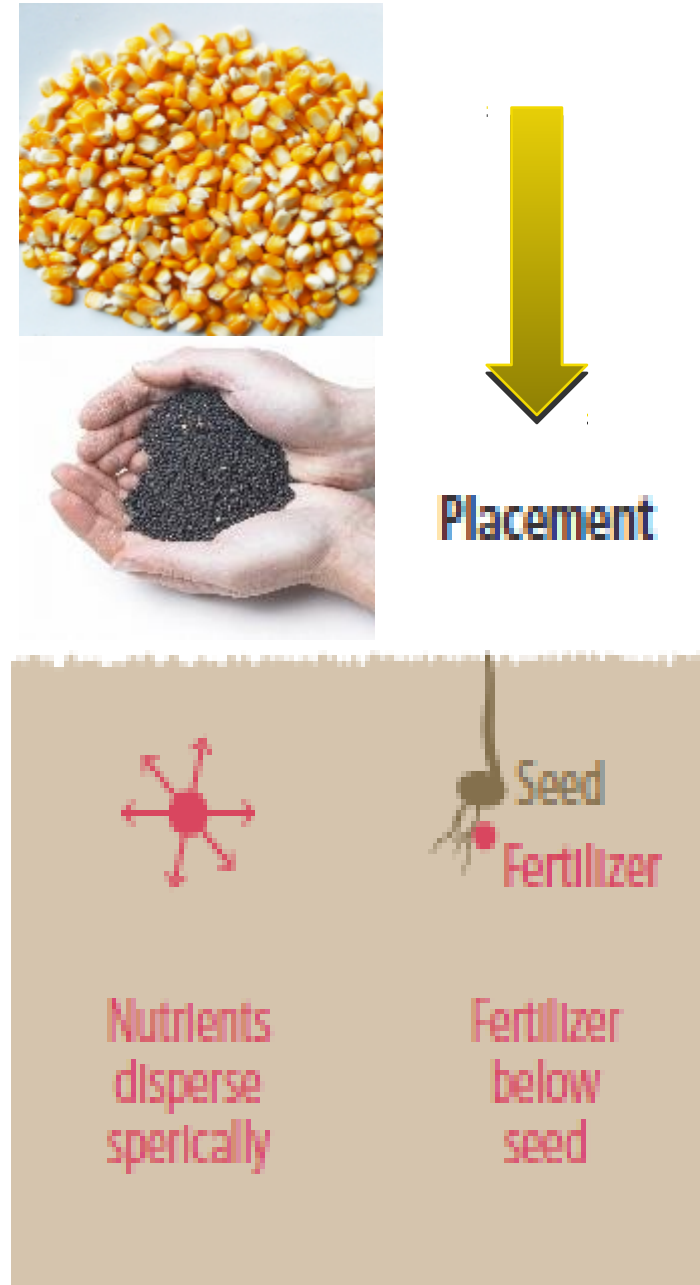
- 🌐 **Precise dosing:** real removal
- 🌐 Nutrient **removed** by the harvest
- 🌐 Nutrient absorbed during the whole vegetative cycle
- 🌐 **uptake** - absorbed from plants during the crop cycle; nutrients must be available
- 🌐 **removal** - removed with the harvested plants
- 🌐 **losses** - leaching



Fertilizers on top the soil



Fertilizers into the soil



Box 3. Types of mineral fertilizers (according to different criteria)

Method of production

- natural (as found in nature or only slightly processed);
- synthetic (manufactured by industrial processes).

Number of nutrients

- single-nutrient or straight fertilizers (whether for major, secondary or micro nutrients);
- multi-nutrient (multiple nutrient) or compound fertilizers, with 2, 3 or more nutrients:

Type of combination

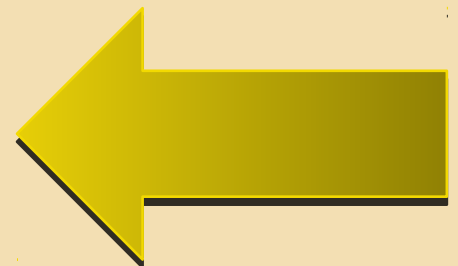
- mixed fertilizers, i.e. a physical mixture of two or more single-nutrient or multi-nutrient fertilizers (for granular products this may comprise a blend of separate granules of the individual ingredients, or granules each containing these ingredients);
- complex fertilizers, in which two or more of the nutrients are chemically combined (e.g. nitrophosphate, ammonium phosphates).

Physical condition

- solid (crystalline, powdered, prilled or granular) of various size ranges;
- liquid (solutions and suspensions);
- gaseous (liquid under pressure, e.g. ammonia).




Mode of action

- quick-acting (water-soluble and immediately available);
- slow-acting (transformation into soluble form required).



IFA definition

Term	Definition	Technology	Example of product
Controlled Release Fertilizer (CRF)	Fertilizer in which nutrient release is <u>controlled, meeting the stated release rate of nutrient and the stated release time at a specified temperature.</u>	Fertilizer coated with a <u>polymeric coating</u>	Osmocote, Agroblen, Sierrablen (Everris) ESN (Agrium), Duration, POLYON (Koch Agronomic Services) Multicote (Haifa) Nutricote (JCAM Agri) Apex (Simplot)
Slow Release Fertilizer (SRF)	Fertilizer, of which, by hydrolysis and/or by biodegradation and/or by <u>limited solubility, the nutrients available to plants is spread over a period of time, when compared to a "reference soluble " product e.g. ammonium sulphate, ammonium nitrate and urea.</u>	Use of <u>less water soluble nutrients and/or compounds which are degraded in soil over time</u>	N-Sure (Tessenderlo Kerley) Sierraform (Everris) Nitroform (Koch Agronomic Services) Floranid (IBDU) (Compo) IBDU (JCAM Agri) Sazolene and Sirflor (Sadepan Chimica) CDU (JCAM Agri)

Water-soluble	Slow-release	Controlled-release
	<p data-bbox="846 404 1083 451">Non-coated</p> 	<p data-bbox="1454 404 1603 451">Coated</p> 
<p data-bbox="239 958 562 1096">Dissolves all at once</p>	<p data-bbox="726 965 1244 1103">Slowly decomposes to soluble</p>	<p data-bbox="1335 965 1760 1103">Nutrients "leak" through coating</p>

- 🌐 Concept of **slow/controlled release** fertilizers: release at gradual rate that permits **maximum uptake** and utilization with **minimization of leaching**, volatilization or excessive growth.
- 🌐 **Fertilizer use efficiency** is higher and regulated by **supply/release** rate of nutrients that is synchronized with **crop demand**.

Why **SLOW/CONTROLLED RELEASE** fertilizers?

- Lower **toxicity**
- Lower **losses** of nutrients - leaching/runoff, evaporation
- Lower environmental **pollution**
- Plants **take up most** of fertilizer nutrients
- Less frequent **application** required
- Only **2 applications/year**; no other fertilizer needed
- Act weeks-months
- Time savings
- Economical

High efficiency of the applied fertilizer

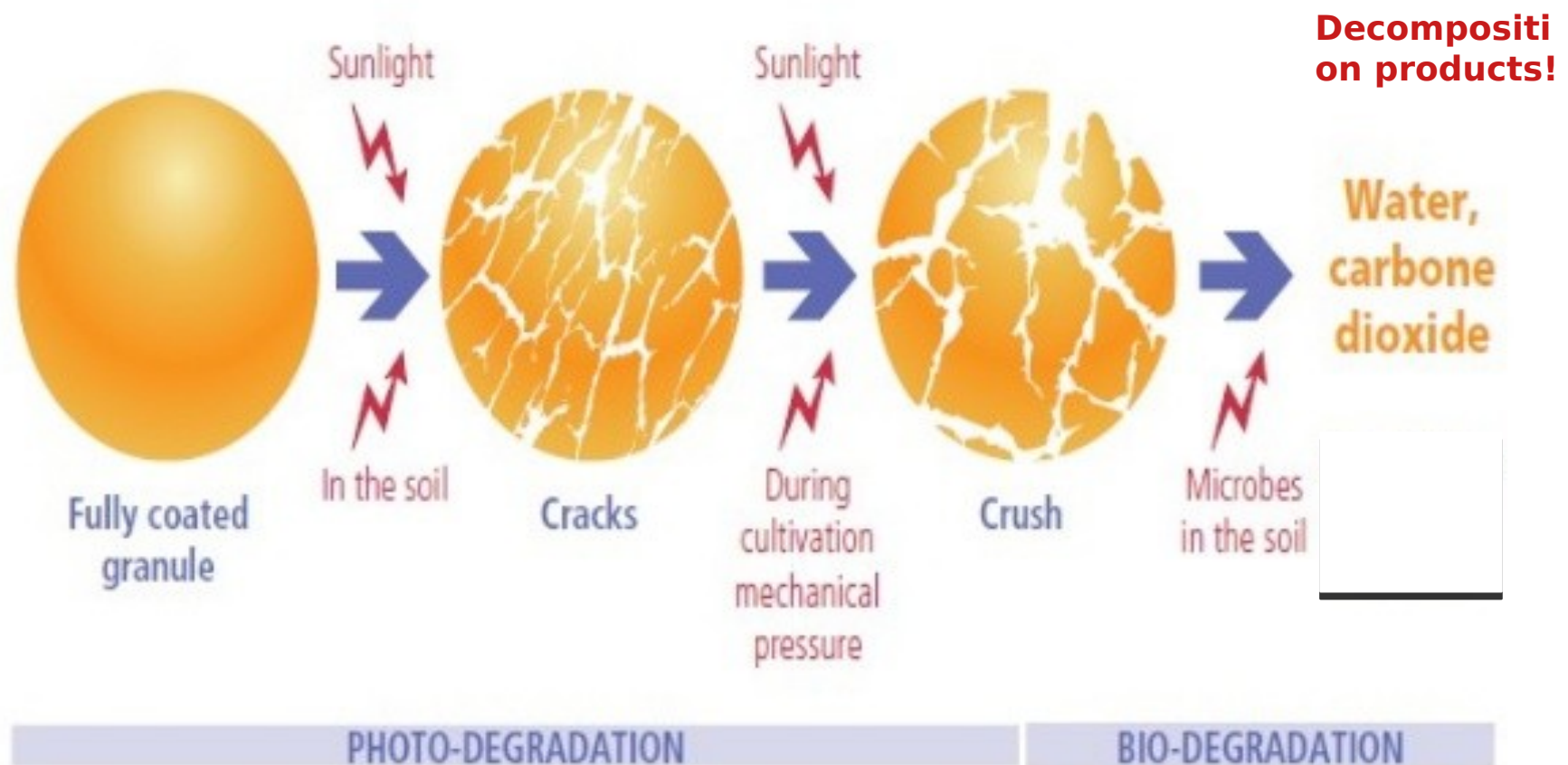
- 🌐 A significant **reduction of nutrient losses** => **NUE** increases.
- 🌐 **Physical barrier or coating** protects the fertilizer granules from dissolving.
- 🌐 Depending on the **thickness** of the coating and its components, the protection can be for a **few days to a few months**.
- 🌐 Conversion pathways are regulated by soil **bacteria**.

Coatings and controlled-release fertilizers

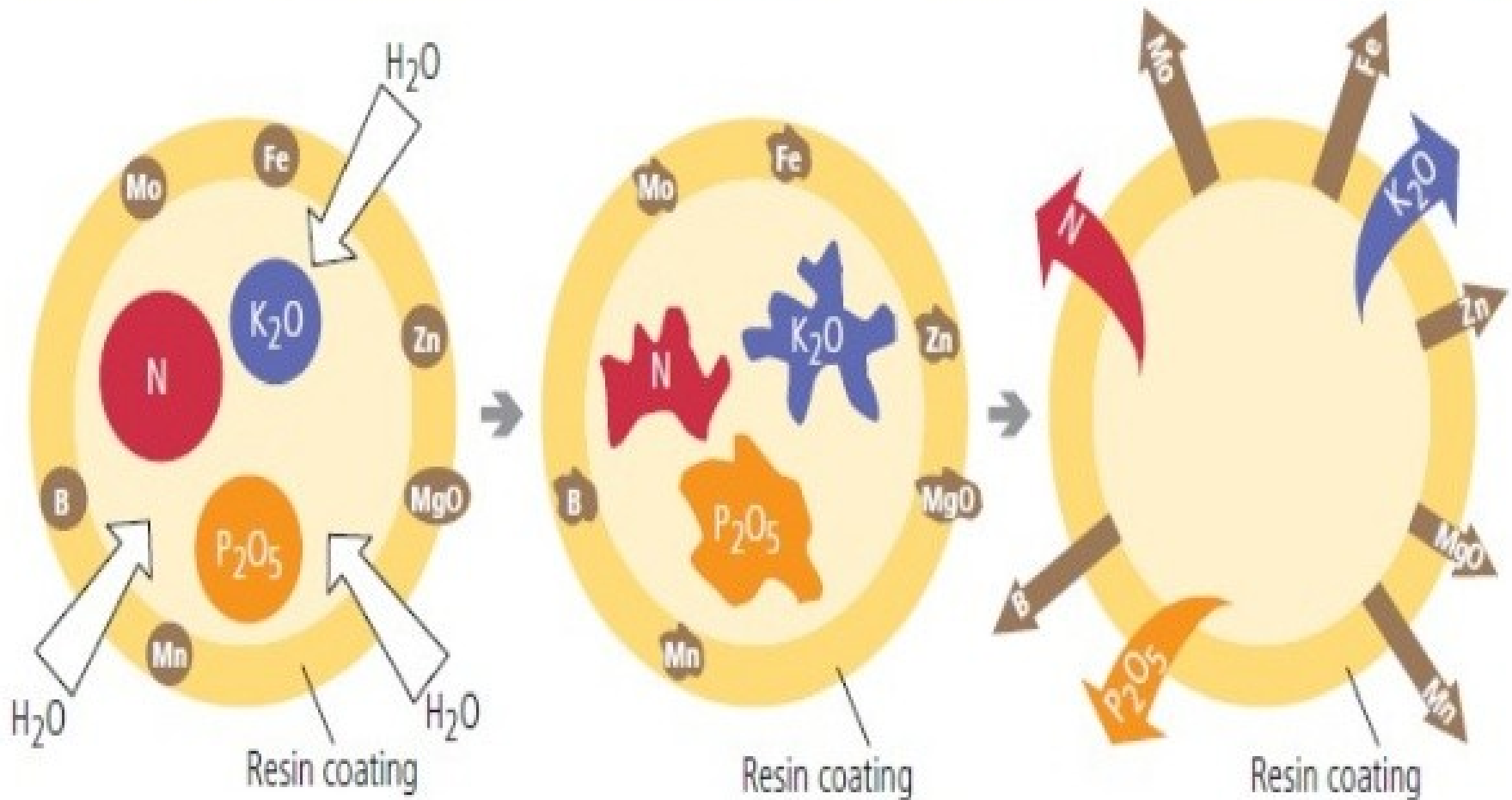
- 🌐 **Coatings on soluble** fertilizers physical barrier.
- 🌐 **Polymer coatings** - to control nutrient release and improve nutrient use efficiency - from a few weeks to several months depending on the coating and the conditions in the soil.
- 🌐 **Coated** - lower losses from the root zone, and releasing the nutrients at a time of uptake required by the plant.
- 🌐 **Used in:**
 - 🌐 vegetable, orchards, ornamental systems;
 - 🌐 possible for cereal grain system

- Polymer-coated fertilizers – more recent technology for controlling nutrient release and reducing losses by leaching
- Different **types of coatings** – plaster of paris, wax, polymer – polyacrylamide – reduces soil erosion
- Environmentally smart coating - **flexible polymer** coating, which controls the **permeability to water** absorption and **dissolved nutrient release**. Applied at different **thicknesses to adjust the rate** at which water can be absorbed and the nutrients dissolved.
- Polymer coated **multi-nutrient fertilizers** can supply all required fertilizer elements

Decomposition model of the coating polymer of CONTROLLED release fertilizer



Mode of action of coated/encapsulated fertilizer



SLOW RELEASE

- Release by diffusion/hydrolysis - depends on **soil parameters**:
 - water content,
 - pH,
 - ionic content,
 - temperature.
- **Insoluble compounds** can be effective fertilizers if **release rate matches plant requirement** in the growth period.
- The mechanism of nutrients release can be different.
- The compounds should have **low solubility in water and high in organic acids (citrate)** and DTPA.

 **Micronutrient SRF:**


 **insoluble oxides**


 **metaphosphates/glassy phosphates/glass beads**

 **liquid polymers**

 **Glassy phosphate is produced by fusing oxides of micronutrients in phosphoric acid at 8000 °C.**

 **Dissolve by slow hydrolysis to release nutrients to soil.**

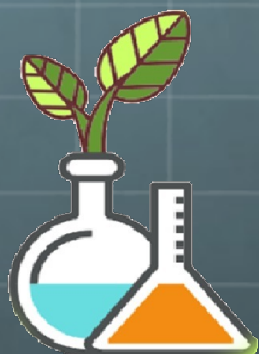
 make the nutrients easier to apply uniformly and also help control the rate of release.

 bind with soil cations and help maintain solubility under some conditions.

Conclusions

- 🌐 Fertilizers can be made more efficient by:
 - 🌐 **slowing the release** of nutrients,
 - 🌐 **inhibiting conversion** to forms that are less stable in the soil,
 - 🌐 or **enhancing availability** of nutrients to plants.
- 🌐 in terms of agronomic, economic and environmental factors.
- 🌐 **Enhancing fertilizer use efficiency**
- 🌐 Adding these control systems usually **adds to the costs**, but may be justified if sufficient enhancement of **nutrient efficiency** and

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